



DETERMINATION OF WATER-SOLUBLE VITAMINS CONTENT IN PASTRY USING THE HPLC METHOD

Saidjahon Saidumarov

Master's Student at the International Institute of Food Technology and
Engineering, Fergana, Uzbekistan
E-mail: saidjahansaidumarov@gmail.com

Barnokhon Sattarova

Associate Professor, Fergana State Technical University, Fergana, Uzbekistan
E-mail: sattarovabarno1967@gmail.com

Abstract

For the extraction of water-soluble vitamins, 1 g of the sample was accurately weighed and placed into a 50 mL conical flask, followed by the addition of 25 mL of 0.1 N hydrochloric acid (HCl) solution. The mixture was subjected to ultrasonic extraction using a GT SONIC-D3 ultrasonic bath (China) at 60°C for 20 minutes. The standard solutions and sample extract were analyzed using a high-performance liquid chromatography (HPLC) system comprising an LC-40D pump, SIL-40 autosampler, and SPD-M40 photodiode array (PDA) detector, operated with LabSolutions software version 6.92, based on the LC-40 Nexera Lite platform.

Keywords: pâté; vitamin B₁₂; vitamin C; vitamin content; sample extract; calorie; vitamins.

Introduction

Pâté is a food product prepared from the meat and offal of domestic animals. In Uzbekistan, the consumption of pâté is not yet widespread. However, pâtés are rich in calories and considered a nutritious food that can be easily prepared at home. In addition, liver pâtés are regarded as dietary foods and are known to assist in the treatment of anemia and improve the functioning of the nervous system [1-3]. Promoting the consumption of pâté products among the population could help partially meet the nutritional and caloric needs of individuals. The main objective of the present study is to highlight the significance of pâté in the food sector and to promote its wider consumption [4-7].

Materials and Equipment Used

Vitamin B₁₂ was obtained from *Rhydburg Pharmaceuticals* (Germany), vitamin C from *Carl Roth GmbH* (Germany), vitamin B₉ from *DSM Nutritional Products GmbH* (Germany), and vitamins B₁, B₂, B₃, B₆, and PP from *BLDPharm* (China). HPLC-grade water, acetonitrile, glacial acetic acid, and analytical-grade sodium hydroxide were used as reagents [8-13].



The determination of water-soluble vitamin contents was performed using a high-performance liquid chromatography (HPLC) system, LC-40 Nexera Lite, manufactured by Shimadzu Corporation (Japan) [1].

Preparation of Standard Solutions

Standard solutions of vitamins C (CAS 50-81-7), B₁ (CAS 59-43-8), B₆ (CAS 58-56-0), B₃ (CAS 59-67-6), B₁₂ (CAS 68-19-9), and PP (CAS 98-92-0) were prepared by dissolving 5 mg of each vitamin in 50 mL of 0.1 N hydrochloric acid (HCl) to achieve a concentration of 100 mg/L. Standard solutions of vitamins B₂ (CAS 83-88-5) and B₉ (CAS 59-30-3) were prepared by dissolving 5 mg of each in 50 mL of 0.025% sodium hydroxide (NaOH) solution. Subsequently, 200 µL aliquots of the B₁, B₆, B₃, B₁₂, and PP vitamin standard solutions were mixed to prepare a combined solution with a concentration of 14.286 mg/L for each vitamin. Further dilutions were prepared to obtain standard solutions of 7.143, 3.571, and 1.786 mg/L.

Similarly, standard solutions of vitamin C were prepared at concentrations of 286, 143, 71.5, and 57.2 mg/L. Distilled water was used for the preparation of the 0 mg/L blank sample for calibration curve construction [14-19].

Preparation of Sample Extract

For the extraction of water-soluble vitamins, 1.0 g of the test sample was weighed and placed into a 50 mL conical flask, followed by the addition of 25 mL of 0.1 N HCl solution. The mixture was subjected to ultrasonic-assisted extraction using a GT SONIC-D3 ultrasonic bath (China) at 60°C for 20 minutes. After extraction, the mixture was cooled to room temperature, filtered, and the final volume was adjusted to 25 mL with distilled water in a volumetric flask. An aliquot of 1.5 mL of the extract was filtered through a 0.22 µm syringe filter into a vial and used for subsequent HPLC analysis [20-23].

Chromatographic Conditions

Determination of Vitamins. The standard solutions and sample extracts were analyzed using a high-performance liquid chromatography (HPLC) system comprising an LC-40D pump, SIL-40 autosampler, and SPD-M40 photodiode array (PDA) detector, operated with LabSolutions software version 6.92, based on the LC-40 Nexera Lite platform (Shimadzu Corporation, Japan) [24-26].

Separation was achieved using a reversed-phase Shim-pack GIST C18 column (150 × 4.6 mm; 5 µm, Shimadzu, Japan) with a gradient mobile phase composed of acetonitrile (solvent A) and 0.25% aqueous acetic acid solution (solvent B), as described in Table 1 [27-29].

The injection volume was set at 10 µL, the flow rate at 0.6 mL/min, and the column oven temperature was maintained at 40°C. Analytical signals (peak areas) for each vitamin were recorded at three different detection wavelengths: 265 nm, 291 nm, and 550 nm (Figures 1–3) [28]. For the determination of vitamin C, a specific 15-minute gradient program was applied (Table 2), and the analytical signal was monitored at 265 nm [30].

Table 1. Gradient Program Used for Vitamin Determination

Time, min.	Acetonitrile (A), %	0.5% acetic acid (B), %
0	0	100
3	0	100
14	20	80
17	50	50
18	0	100
25	Finish	

Table 2. Mobile phase gradient program for vitamin C quantification.

Time, min.	Acetonitrile (A), %	0.5% acetic acid (B), %
0	0	100
2	0	100
6	50	50
6,01	0	100
15	Finish	

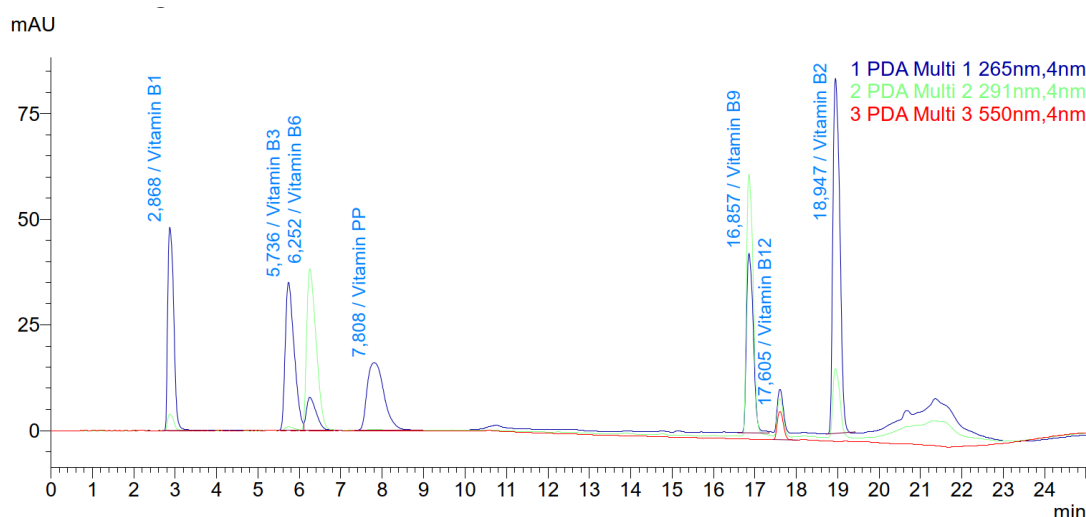


Figure 1. Chromatogram of a standard solution of vitamins.

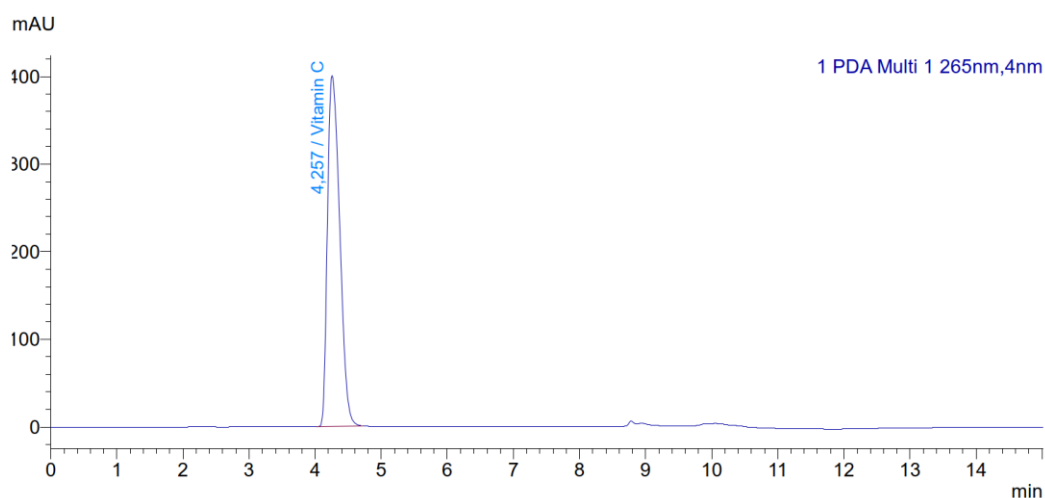


Figure 2. Chromatogram of a vitamin C standard solution

Results

Determination of vitamins in the sample extract. A chromatogram of the sample extract (Figures 3-4) was obtained and based on the results, the amounts of vitamins in 100 g of the sample were calculated using the following formula and are presented in Table 3.

$$X = \frac{C_{vit} \cdot V_{extract}}{m_{sample}} \cdot 100 \text{ g}$$

Here,

X – the amount of vitamins in 100 grams of the sample, expressed in mg;

C_{vit} – the concentration of the vitamin in the extract determined by the HPLC method, expressed in mg/L;

$V_{extract}$ – the volume of the sample extract, expressed in liters (L);

m_{sample} – the mass of the sample taken for extraction, expressed in grams

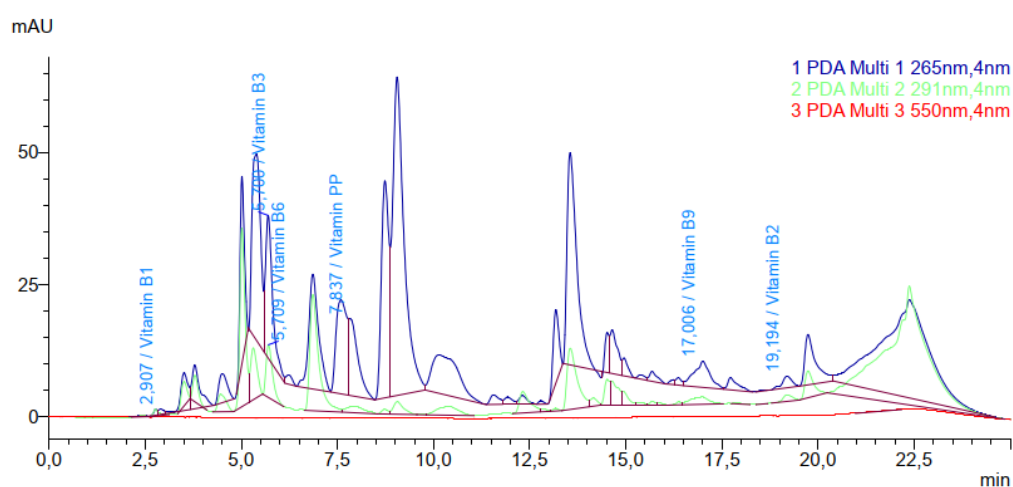


Figure 3. Chromatogram of the determination of vitamins in the sample extract.

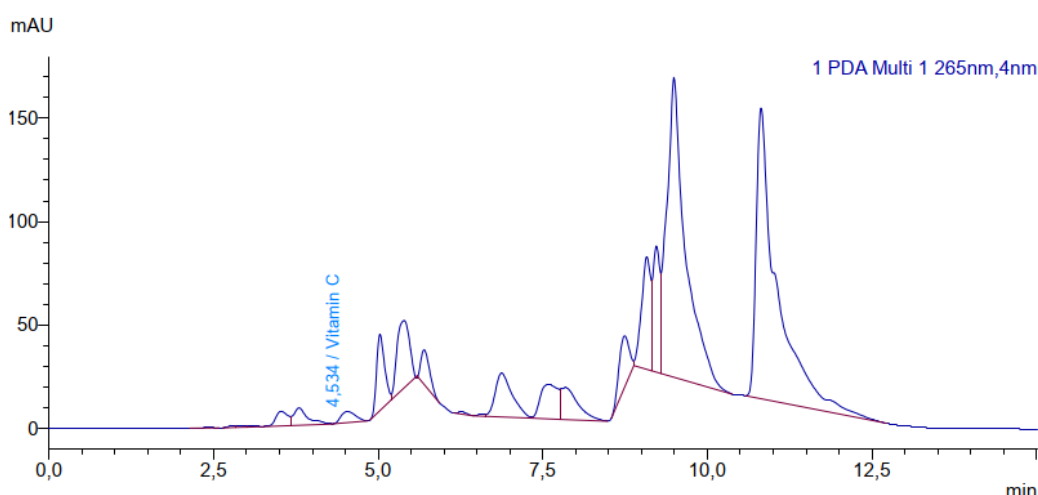


Figure 4. Chromatogram of vitamin C in the sample extract.

Table 3. Amount of vitamins in the extract and retention times.

Vitamin	Capture time, sec	Concentration, mg/l	Amount in 100 g of sample, mg
Vitamin B ₁	2,907	0,343	0,858
Vitamin B ₃	5,7	9,251	23,128
Vitamin PP	7,837	7,403	18,508
Vitamin B ₉	17,006	4,186	10,465
Vitamin B ₂	19,194	0,839	2,098
Vitamin B ₆	5,709	2,752	6,880
Vitamin B ₁₂	Not specified	0	0,000
Vitamin C	4,534	4,271	10,678

Conclusions

In this study, a high-performance liquid chromatography (HPLC) method was successfully developed and applied for the determination of water-soluble vitamins in pâté samples. The ultrasonic-assisted extraction with 0.1 N hydrochloric acid at 60°C, combined with optimized chromatographic conditions using a reversed-phase C18 column, allowed for the accurate identification and quantification of vitamins B₁, B₂, B₃, B₆, B₉, B₁₂, PP, and C.

The analysis demonstrated that certain vitamins, particularly vitamin B₆ and vitamin C, were present in notable concentrations, while vitamin B₁₂ was either undetected or found at trace levels. The method exhibited high sensitivity, repeatability, and precision, making it a reliable approach for the assessment of water-soluble vitamins in complex food matrices such as pâté.

Given the nutritional significance of these vitamins, promoting the consumption of pâté products enriched with essential micronutrients could contribute to improving the dietary intake and overall health of the population. Future research should focus on validating the method across different types of pâté formulations and studying the stability of water-soluble vitamins during storage and processing.

References

- Asqarov, I. R., Abdullayev, S. S. O., Mamatqulova, S. A., Abdulloyev, O. S., & Abdulloyev, S. X. (2024). Development of a method for determining the content of water-soluble vitamins using HPLC (on the example of cilantro). *Fergana State University Scientific Journal*, 30(5), 61. <https://journal.fdu.uz/index.php/sjfsu/article/view/4679>
- Shimadzu Corporation. (2020). LC-40 Nexera Lite High Performance Liquid Chromatography System: Operation Manual.
- Singleton, V. L., Orthofer, R., & Lamuela-Raventós, R. M. (1999). Analysis of total phenols and other oxidation substrates and antioxidants by means of Folin–Ciocalteu reagent. *Methods in Enzymology*, 299, 152–178.
- Macheix, J. J., Fleuriet, A., & Billot, J. (1990). *Fruit Phenolics: Chemistry, Phytochemistry and Modulation of Quality*. CRC Press.
- Shodiev, D. A. U., & Najmitdinova, G. K. K. A. (2021). Specific aspects of food production. *Universum: Technical Sciences*, 3–2(84), 91–94.

6. Dilshodjon, S., & Hojiali, Q. (2022). Importance of food colorings in the food industry. *Universum: Technical Sciences*, 11–8(104), 23–25.
7. Shodiev, D. A. (2022). The significance of biological quantities of microelements in plants. *Formation of Psychology and Pedagogy as Interdisciplinary Sciences*, 1(9), 297–301.
8. Shodiev, D. A. U., & Kurbanov, K. A. U. (2022). Prospects for the use of food additives in the food industry. *Universum: Technical Sciences*, 5–7(98), 24–26.
9. Shodiev, D. A. U., & Rasulova, U. N. K. (2022). The importance of amaranth oil in medicine. *Universum: Technical Sciences*, 1–2(94), 69–72.
10. Shodiev, D., Haqiqatkxon, D., & Zulaykho, A. (2021). Useful properties of the amaranth plant. *ResearchJet Journal of Analysis and Inventions*, 2(11), 1–4.
11. Shodiev, D., & Hojiali, Q. (2021). Medicinal properties of amaranth oil in the food industry. *Interdisciplinary Conference of Young Scholars in Social Sciences*, 205–208.
12. Shodiev, D. A., & Najmitdinova, G. K. (2021). Food additives and their significance. *Universum: Technical Sciences*, 10–3(91), 30–32.
13. Kholdarov, D. M., Shodiev, D. A., & Rayimberdieva, G. G. (2018). Geochemistry of microelements in elementary landscapes of the desert zone. *Actual Problems of Modern Science*, (3), 77–84.
14. Kholdarov, D., et al. (2021). General characteristics and mechanical composition of saline meadow saz soils. *Conference Proceedings*.
15. Dilshodjon, S., & Hojiali, Q. (2022). Nutritional value of food supplements and their impact on the body. *Universum: Technical Sciences*, 12–7(105), 32–35.
16. Dilshod, S., Hojiali, Q., & Gulbakhroy, S. (2023). Biological properties of the medicinal plant amaranth and its significance in the food industry. *Universum: Technical Sciences*, 3–5(108), 19–21.
17. Dilshod, S., & Hojiali, Q. (2023). Chemical analysis of amaranth oil and its beneficial properties. *Universum: Technical Sciences*, 2–6(107), 29–30.
18. Dilshod, S., Hojiali, Q., & Mohidil, A. (2023). The value of compounds that change the color of food raw materials and finished products. *Universum: Technical Sciences*, 4–7(109), 52–54.
19. Dilshod, S., Hojiali, Q., & Mohidil, A. (2023). Features of the use of valuable natural food dyes in the food industry. *Universum: Technical Sciences*, 5–7(110), 56–58.
20. Shodiev, D. A., & Abduvalieva, M. A. (2023). Biological research of local medicinal plants used in animal feeding in agriculture. *Modern Trends in Biology: Problems and Solutions*, 1(4), 687–689.
21. Shodiev, D., & Abduvalieva, M. (2023). The value of amaranth food additives in the food industry. *Texas Journal of Agriculture and Biological Sciences*, 23, 67–71.
22. Ergashov, A. A., & Abrolov, A. A. (2024). Adsorbents used in industry and challenges in their application. *Research and Implementation*, 2(7), 26–31.
23. Kodirov, Z. Z., & Ahmadjonovich, A. A. (2023). Research and control measures of powdery mildew (oidium) diseases in vine fruit production. *European Journal of Emerging Technology and Discoveries*, 1(2), 86–92.
24. Adahamjonovich, A. A. (2022). Diarrhea and healing function from watermelon seed. *International Journal of Advance Scientific Research*, 2(5), 84–89.





25. Nabievna, S. B., & Adxamjonovich, A. A. (2021). The chemical composition and properties of chicken meat. *Innovative Technologica: Methodical Research Journal*, 2(10), 25–28.
26. Mahammadjon, Q., & Anvar, A. (2021). Bioazot-N biopreparations in agriculture. *Innovative Technologica: Methodical Research Journal*, 2(11), 101–105.
27. Madaliyev, T. A., Goppirjonovich, Q. M., & Abrolov, A. A. (2020). Bioprospecting of exopolysaccharide-producing bacteria from various natural ecosystems for biopolymer synthesis from bardy. *Universum: Chemistry and Biology*, 12–1(78), 6–9.
28. Qosimov, M. G., Madaliyev, T. A., & Abrolov, A. A. (2019). Improving the quality of grains grown in the conditions of the Fergana region. *Internauka*, 40–2, 28–30.
29. Ibragimov, A. A., et al. (2019). On the prospects of organizing the fishing industry in Uzbekistan and the development of fish farming in the reservoirs of the Fergana Valley. *Universum: Technical Sciences*, 12–3(69), 21–23.
30. Kurbanov, J. Kh., et al. (2019). Heat exchange intensity during heating of $\text{NH}_2\text{COONH}_4$ solution in a heat exchanger with highly efficient pipes. *Universum: Technical Sciences*, 12–2(69), 24–27.